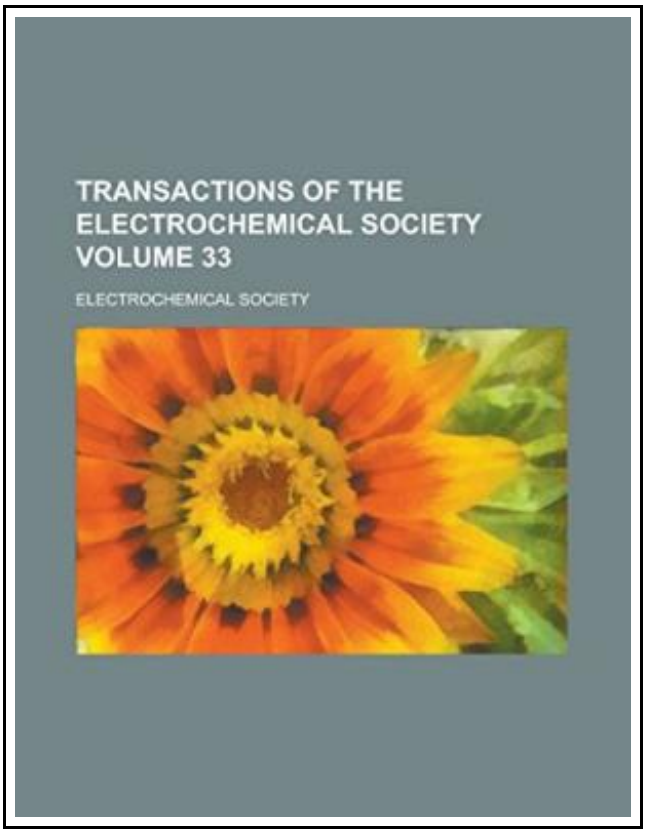
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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1919 edition. Excerpt: .constant, then made a determination of the current efficiency at that point, giving us two valuable data for each iron content; namely the ratio ferric to ferrous iron, and the efficiency at that ratio. If he had followed it out to that extent, this table would have been of more value, and I think also would have eliminated those somewhat erratic figures in the last column which do not plot to smooth curves. Referring to page 141, conclusion No. 7; of course that resolves itself on the presence of iron, but he should say the effect of the presence of ferric and ferrous iron, and treat the ferric and the ferrous iron as two separate substances. C. G. Fink: This problem of the effect of iron on the electrolysis of copper is rather complex and there are a number of us who are busily engaged on experiments along this line, but there is nothing definitely established as yet as far as I know. Perhaps some other gentleman present can tell us a little more. J. L. Yardley3: According to my understanding at the Electrolytic Copper Plant at A jo, Arizona, the solution as it enters the cell room, contains a combination of ferrous and ferric iron, the ferrous iron preponderating. The iron in the solution is changed as it passes through the cell room and, as it leaves, the ferric iron preponderates. Before the solution returns to the cell room it goes through reduction...

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